

# Virial expansion for charged colloids and electrolytes

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**Abstract.** Using a field-theoretic approach, we derive the first few coefficients of the exact low-density (“virial”) expansion of a binary mixture of positively and negatively charged hard spheres (two-component hard-core plasma, TCPHC). Our calculations are nonperturbative with respect to the diameters  $d_+$  and  $d_-$  and charge valences  $q_+$  and  $q_-$  of positive and negative ions. Consequently, our closed-form expressions for the coefficients of the free energy and activity can be used to treat dilute salt solutions, where typically  $d_+ \sim d_-$  and  $q_+ \sim q_-$ , as well as colloidal suspensions, where the difference in size and valence between macroions and counterions can be very large. We show how to map the TCPHC on a one-component hard-core plasma (OCPHC) in the colloidal limit of large size and valence ratio, in which case the counterions effectively form a neutralizing background. A sizable discrepancy with the standard OCPHC with uniform, rigid background is detected, which can be traced back to the fact that the counterions cannot penetrate the colloids. For the case of electrolyte solutions, we show how to obtain the cationic and anionic radii as independent parameters from experimental data for the activity coefficient.

**PACS.** 82.70.-y Disperse systems; complex fluids – 52.25.-b Plasma properties – 61.20.Qg Structure of associated liquids: electrolytes, molten salts, etc.

## 1 Introduction

The two-component hard-core plasma (TCPHC) has been used for a long time as an idealized model for electrolyte solutions. In this model, also known as “primitive model”, the ions are spherical particles that interact with each other *via* the Coulomb potential and a hard-core potential, which avoids the collapse of oppositely charged particles onto each other. In the general asymmetric case, the positive ions have a charge  $q_+e$  (where  $e$  is the elementary charge) and ionic diameter  $d_+$ , while the negative ions have a charge  $-q_-e$  and diameter  $d_-$ . The particles are immersed in a structureless solvent whose presence is felt only through the value of the dielectric constant of the medium, and the system is (globally) electroneutral.

The TCPHC in this formulation is a gross simplification of real systems. When two ions are at distances of the order of the size of the solvent molecules, the assumption of a continuum solvent breaks down, giving rise to so-called solvation forces [1]. Such effects also lead to non-local contributions to the dielectric constant. Since the

dielectric constant of ions or colloids typically differs from the surrounding aqueous medium, one also expects dispersion forces to act between these particles. All these effects, which contribute to “ion-specific” effects [2], and are possibly more relevant than previously thought [3], are not accounted for in the TCPHC, which treats the solvent — usually water — as a structureless medium within which the charged particles are embedded, neglecting the molecular arrangement that occurs around the ions.

Nevertheless, even with such simplifications, the TCPHC is far from being amenable to an exact treatment. A better understanding of this model is a necessary step if one wishes to develop more realistic approaches to charged systems. In this article, we turn our attention to the low-density, or virial, expansion of the TCPHC. Since we have exact results for the thermodynamic properties at low concentrations using field-theoretic methods, we can obtain useful information on dilute systems with otherwise arbitrary hard-core radii and charge valences (like colloidal suspensions). In particular, we can test various approximations to treat the TCPHC, like the mapping of a colloidal solution on an effective one-component plasma.

Due to the long-range character of the Coulomb potential, it is not easy to obtain the thermodynamic behavior of the TCPHC through the usual methods of statistical mechanics. For example, it can be shown [4–6] that the straightforward application of the cluster expansion to

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the TCPHC leads to divergent virial coefficients. Mayer [4] proposed a solution to this problem through an infinite resummation of the cluster diagrams, carried out such that the divergent contributions to the virial expansion are canceled. With this, he was able to obtain explicitly the first term in the virial (or low-density) expansion that goes beyond the ideal gas, which turns out to be the well-known Debye-Hückel limiting law [7]. Haga [8] carried the expansion further and went up to order 5/2 in the ionic density. More or less at the same time, Edwards [9] also obtained the virial expansion of the TCPHC by mixing cluster expansion and field theory. In both cases only equally-sized ions were considered.

The aforementioned methods typically depend on drawing, counting and recollecting the cluster diagrams which give finite contributions to the expansion up to the desired order in the density. This can be quite a formidable task, and unfortunately it is easy to “forget” diagrams that are relevant to the series (see for instance the comment on pp. 222–223 of Ref. [5]). Also, the generalization to ions with different sizes is rather complicated [5]. Besides, the final results are typically not obtained in closed form, *i.e.*, the final expressions depend on infinite sums that usually have to be evaluated numerically, which reflects the infinite diagrammatic resummation.

We generalize here a novel field-theoretic technique [10], introduced for the symmetric TCPHC ( $q_+ = q_-$  and  $d_+ = d_-$ ). We obtain the exact low-density expansion of the *asymmetric* (both in size and charge) TCPHC. This method does not use the cluster expansion (and resummation) and yields analytic, closed-form results. We go up to order 5/2 in the volume fraction of a system where the sizes and the charge valences of positive and negative ions are unconstrained, that is, the results we obtain can be applied, without modifications, to both electrolyte solutions (where anions and cations have approximately the same size and valence) and to colloidal suspensions (where the macro- and counterions have sizes and valences that can be different by orders of magnitude). In the colloidal limit, we demonstrate how to map the TCPHC on an effective one-component hard-core plasma (OCPHC) and obtain the corrections due to the exclusion of the background (*i.e.* the counterions) from the colloidal particles. For electrolytes, we obtain effective ionic sizes in solution, or thermodynamic diameters, from experimental data for the mean activity. The method developed here allows, in principle, to measure the diameters of cations and anions as *independent* variables, given that the range of experimental data extends to low enough densities such that the expansion used here is valid. It should be noted that the diameters obtained in this way are fundamentally different from the hydrodynamic diameters.

This article is organized as follows. In Section 2 we describe in detail the steps that lead to the low-density expansion. Readers which are not interested in this derivation can go directly to Section 3, where we apply the expressions obtained to two particular problems, *viz.*, (i) colloids, where one of the charged species is much smaller and much less charged than the other one and (ii) the mean

activity coefficient (which is related to the exponential of the chemical potential) of electrolyte solutions, which is available from experiments and can be compared to our results. Finally, Section 4 contains some concluding remarks.

## 2 The method

We begin our calculation by assuming a system with  $N_+$  positively charged particles with charge valence  $q_+$  and diameter  $d_+$ , and  $N_-$  negatively charged particles with charge valence  $q_-$  and diameter  $d_-$ . Global electroneutrality of the system will be imposed at a later stage of the calculation. The canonical partition function  $\mathcal{Z}$  is given by

$$\mathcal{Z} = \frac{1}{N_+!N_-!} \int \prod_{i=1}^{N_+} \frac{d\mathbf{r}_i^{(+)}}{\lambda_{T,+}^3} \prod_{j=1}^{N_-} \frac{d\mathbf{r}_j^{(-)}}{\lambda_{T,-}^3} \exp\left(-\frac{\mathcal{H}}{k_B T}\right) \quad (1)$$

where  $\lambda_{T,+}$  and  $\lambda_{T,-}$  are the thermal wavelengths, and  $\mathbf{r}_i^{(+)}$  and  $\mathbf{r}_i^{(-)}$  are the positions of positively and negatively charged particles. The Hamiltonian  $\mathcal{H}$  is given by

$$\begin{aligned} \frac{\mathcal{H}}{k_B T} = & -E_{\text{self}} + \frac{1}{2} \sum_{\alpha,\beta=+,-} \int d\mathbf{r}d\mathbf{r}' \hat{\rho}_\alpha(\mathbf{r}) \omega_{\alpha\beta}(\mathbf{r}-\mathbf{r}') \hat{\rho}_\beta(\mathbf{r}') \\ & + \frac{1}{2} \int d\mathbf{r}d\mathbf{r}' [\hat{\rho}_+(\mathbf{r}) - \hat{\rho}_-(\mathbf{r})] v_c(\mathbf{r}-\mathbf{r}') [\hat{\rho}_+(\mathbf{r}') - \hat{\rho}_-(\mathbf{r}')] \end{aligned} \quad (2)$$

where the charge-density operators of the ions are defined as

$$\begin{aligned} \hat{\rho}_+(\mathbf{r}) &= q_+ \sum_{i=1}^{N_+} \delta(\mathbf{r} - \mathbf{r}_i^{(+)}) \\ \hat{\rho}_-(\mathbf{r}) &= q_- \sum_{i=1}^{N_-} \delta(\mathbf{r} - \mathbf{r}_i^{(-)}) \end{aligned} \quad (3)$$

and  $\delta(\mathbf{r} - \mathbf{r}')$  is the Dirac delta function. The indices  $\alpha$  and  $\beta$  in equation (2) stand for + and -, and the sum over  $\alpha$  and  $\beta$  in equation (2) runs over all possible permutations (*viz.* ++, --, +- and -+), *i.e.*, we consider a different short-ranged potential  $\omega_{\alpha\beta}$  for each combination. The Coulomb potential is given by  $v_c(\mathbf{r}) = \ell_B/r$ , where  $\ell_B \equiv e^2/(4\pi\epsilon k_B T)$  is the Bjerrum length, defined as the distance at which the electrostatic energy between two elementary charges equals the thermal energy  $k_B T$ . Finally,  $E_{\text{self}}$  is the self-energy of the system

$$E_{\text{self}} = \frac{N_+ q_+^2}{2} [\omega_{++}(0) + v_c(0)] + \frac{N_- q_-^2}{2} [\omega_{--}(0) + v_c(0)] \quad (4)$$

which cancels the diagonal terms in equation (2).

We proceed by applying the Hubbard-Stratonovich transformation, the essence of which is given by

$$\begin{aligned} e^{-\frac{1}{2} \int d\mathbf{r}d\mathbf{r}' \hat{\rho}(\mathbf{r}) v(\mathbf{r}-\mathbf{r}') \hat{\rho}(\mathbf{r}')} = \\ \frac{\int \mathcal{D}\phi e^{-\frac{1}{2} \int d\mathbf{r}d\mathbf{r}' \phi(\mathbf{r}) v^{-1}(\mathbf{r}-\mathbf{r}') \phi(\mathbf{r}') - i \int d\mathbf{r} \phi(\mathbf{r}) \hat{\rho}(\mathbf{r})}}{\int \mathcal{D}\phi e^{-\frac{1}{2} \int d\mathbf{r}d\mathbf{r}' \phi(\mathbf{r}) v^{-1}(\mathbf{r}-\mathbf{r}') \phi(\mathbf{r}')}} \end{aligned} \quad (5)$$

where  $v(\mathbf{r})$  is some general potential and  $\int \mathcal{D}\phi$  denotes a path integral over the fluctuating field  $\phi$ . While this transformation can be used without problems when  $v(\mathbf{r})$  is the Coulomb potential, for a short-ranged potential this can be more problematic: for instance, a hard-core potential does not even have a well-defined inverse function. We will anyway take this formal step for the short-ranged potential, and, as we will see later, the way we handle the resulting expressions leads to finite (and consistent) results, *viz.*, the virial coefficients [11].

Applying equation (5) to equation (1) we obtain the partition function in field-theoretic form

$$\mathcal{Z} = \int \frac{\mathcal{D}\psi_+ \mathcal{D}\psi_- \mathcal{D}\phi}{\mathcal{Z}_\psi \mathcal{Z}_\phi} e^{-\bar{\mathcal{H}}_0} W_+ W_- \quad (6)$$

with the action

$$\begin{aligned} \bar{\mathcal{H}}_0 = \frac{1}{2} \sum_{\alpha, \beta=+, -} \int d\mathbf{r} d\mathbf{r}' \psi_\alpha(\mathbf{r}) \omega_{\alpha\beta}^{-1}(\mathbf{r} - \mathbf{r}') \psi_\beta(\mathbf{r}') \\ + \frac{1}{2} \int d\mathbf{r} d\mathbf{r}' \phi(\mathbf{r}) v_c^{-1}(\mathbf{r} - \mathbf{r}') \phi(\mathbf{r}'). \quad (7) \end{aligned}$$

The inverse potentials are formally defined as the solution of the equation

$$\sum_{\beta=+, -} \int d\mathbf{r}' \omega_{\alpha\beta}(\mathbf{r} - \mathbf{r}') \omega_{\beta\gamma}^{-1}(\mathbf{r}' - \mathbf{r}'') = \delta_{\alpha\gamma} \delta(\mathbf{r} - \mathbf{r}'') \quad (8)$$

( $\delta_{\alpha\gamma}$  is the Kronecker delta) and

$$\int d\mathbf{r}' v_c(\mathbf{r} - \mathbf{r}') v_c^{-1}(\mathbf{r}' - \mathbf{r}'') = \delta(\mathbf{r} - \mathbf{r}''). \quad (9)$$

For the Coulomb potential,  $v_c^{-1}(\mathbf{r}) = -\nabla^2 \delta(\mathbf{r}) / 4\pi\ell_B$ . We also define

$$\begin{aligned} W_\alpha = \frac{1}{N_\alpha!} \left[ e^{q_\alpha^2 [\omega_{\alpha\alpha}(0) + v_c(0)] / 2} \right. \\ \left. \times \int \frac{d\mathbf{r}}{\lambda_{T,\alpha}^3} e^{-iq_\alpha [\psi_\alpha(\mathbf{r}) + \alpha \phi(\mathbf{r})]} \right]^{N_\alpha} \quad (10) \end{aligned}$$

and the normalization factors

$$\mathcal{Z}_\psi = \int \mathcal{D}\psi_+ \mathcal{D}\psi_- e^{-\frac{1}{2} \sum_{\alpha\beta} \int d\mathbf{r} d\mathbf{r}' \psi_\alpha(\mathbf{r}) \omega_{\alpha\beta}^{-1}(\mathbf{r} - \mathbf{r}') \psi_\beta(\mathbf{r}')} \quad (11)$$

and

$$\mathcal{Z}_\phi = \int \mathcal{D}\phi e^{-\frac{1}{2} \int d\mathbf{r} d\mathbf{r}' \phi(\mathbf{r}) v_c^{-1}(\mathbf{r} - \mathbf{r}') \phi(\mathbf{r}')} \quad (12)$$

In order to make the calculations simpler we use the grand-canonical ensemble. This is achieved through the transformation

$$\mathcal{Q} = \sum_{N_+, N_- = 0}^{\infty} \Lambda_+^{N_+} \Lambda_-^{N_-} \mathcal{Z}, \quad (13)$$

where  $\Lambda_+$  and  $\Lambda_-$  are, respectively, the fugacities (exponential of the chemical potential) of the positively and negatively charged particles. We perform the sum over  $N_+$  and  $N_-$  without constraints, *i.e.*, without imposing the electroneutrality condition  $q_+ N_+ = q_- N_-$ . Imposing this condition before going to the grand-canonical ensemble makes the calculations much more difficult. Later, electroneutrality will be imposed order-by-order in the low-density expansion in a consistent way, any infinities arising from the non-neutrality of the system will then be automatically canceled.

In its full form, the grand-canonical partition function  $\mathcal{Q}$  reads

$$\begin{aligned} \mathcal{Q} = \int \frac{\mathcal{D}\psi_+ \mathcal{D}\psi_- \mathcal{D}\phi}{\mathcal{Z}_\psi \mathcal{Z}_\phi} \exp \left( \frac{\Lambda_+}{\lambda_{T,+}^3} \int d\mathbf{r} h_+(\mathbf{r}) e^{-iq_+ \phi(\mathbf{r})} \right. \\ \left. + \frac{\Lambda_-}{\lambda_{T,-}^3} \int d\mathbf{r} h_-(\mathbf{r}) e^{iq_- \phi(\mathbf{r})} - \bar{\mathcal{H}}_0 \right), \quad (14) \end{aligned}$$

where  $\bar{\mathcal{H}}_0$  is given in equation (7). We defined the local non-linear operator

$$h_\alpha(\mathbf{r}) \equiv \exp \left( \frac{q_\alpha^2}{2} [\omega_{\alpha\alpha}(0) + v_c(0)] - iq_\alpha \psi_\alpha(\mathbf{r}) \right) \quad (15)$$

(as before,  $\alpha$  stands for both  $+$  and  $-$ ). At this point we rescale the fugacities such that  $\lambda_+ = \Lambda_+ / \lambda_{T,+}^3$  and  $\lambda_- = \Lambda_- / \lambda_{T,-}^3$ , *i.e.*, the fugacities have from now on dimensions of inverse volume.

Introducing the Debye-Hückel propagator,

$$v_{\text{DH}}^{-1}(\mathbf{r} - \mathbf{r}') = v_c^{-1}(\mathbf{r} - \mathbf{r}') + I_2 \delta(\mathbf{r} - \mathbf{r}') \quad (16)$$

with the ionic strength

$$I_2 = q_+^2 \lambda_+ + q_-^2 \lambda_- \quad (17)$$

and after some algebraic manipulations, we finally obtain the grand-canonical free energy density. It is defined through  $g \equiv -\ln(\mathcal{Q})/V$ , and reads

$$\begin{aligned} g = -\lambda_+ - \lambda_- - \frac{1}{2} I_2 v_c(0) - \frac{1}{V} \ln \left( \frac{\mathcal{Z}_{\text{DH}}}{\mathcal{Z}_\phi} \right) \\ - \frac{1}{V} \ln \left\langle e^{\lambda_+ \int d\mathbf{r} Q_+(\mathbf{r}) + \lambda_- \int d\mathbf{r} Q_-(\mathbf{r})} \right\rangle, \quad (18) \end{aligned}$$

where  $V$  is the volume of the system and the brackets  $\langle \dots \rangle$  denote averages over the fluctuating fields  $\phi$  and  $\psi_\alpha$  with the propagators  $\omega_{\alpha\beta}^{-1}$  and  $v_{\text{DH}}^{-1}$ .  $\mathcal{Z}_{\text{DH}}$  is defined as

$$\mathcal{Z}_{\text{DH}} = \int \mathcal{D}\phi e^{-\frac{1}{2} \int d\mathbf{r} d\mathbf{r}' \phi(\mathbf{r}) v_{\text{DH}}^{-1}(\mathbf{r} - \mathbf{r}') \phi(\mathbf{r}')} \quad (19)$$

and the functions  $Q_+(\mathbf{r})$  and  $Q_-(\mathbf{r})$  are defined by

$$Q_\alpha(\mathbf{r}) = h_\alpha(\mathbf{r}) e^{-i\alpha q_\alpha \phi(\mathbf{r})} - 1 + \frac{1}{2} q_\alpha^2 \phi^2(\mathbf{r}) - \frac{1}{2} q_\alpha^2 v_c(0). \quad (20)$$

In the preceding steps we obtained the exact expression for the grand-canonical free energy density  $g$ , still without imposing electroneutrality. In order to obtain the low-density expansion of the free energy, we now (i) expand  $g$  in powers of  $\lambda_+$  and  $\lambda_-$  (up to a order 5/2), (ii) calculate the concentrations of positive and negative particles and impose electroneutrality *consistently*, order-by-order, and (iii) make a Legendre transformation back to the canonical ensemble.

## 2.1 Expansion in powers of the fugacities

We start the expansion of  $g$  by noting that the Fourier transform of the Coulomb potential is  $\tilde{v}_c(\mathbf{k}) = 4\pi\ell_B/k^2$ . Using this, one is able to express  $v_c(0)$  as

$$v_c(0) = \int \frac{d\mathbf{k}}{(2\pi)^3} \frac{4\pi\ell_B}{k^2}. \quad (21)$$

It is easy to show that

$$\frac{1}{V} \ln \left( \frac{\mathcal{Z}_{\text{DH}}}{\mathcal{Z}_\phi} \right) = -\frac{1}{2} \int \frac{d\mathbf{k}}{(2\pi)^3} \ln \left( 1 + \frac{4\pi\ell_B I_2}{k^2} \right). \quad (22)$$

Using equations (21, 22), one finds

$$\frac{1}{2} I_2 v_c(0) + \frac{1}{V} \ln \left( \frac{\mathcal{Z}_{\text{DH}}}{\mathcal{Z}_\phi} \right) = \frac{1}{12\pi} [4\pi\ell_B I_2]^{3/2}. \quad (23)$$

Introducing the dimensionless quantity

$$\Delta v_0 = \sqrt{4\pi\ell_B^3 I_2} \quad (24)$$

and expanding the last term in the rhs of equation (18) in cumulants of  $Q_+(\mathbf{r})$  and  $Q_-(\mathbf{r})$ , one obtains the low-fugacity expansion

$$g = -\lambda_+ - \lambda_- - \frac{\Delta v_0^3}{12\pi\ell_B^3} - \lambda_+ Z_1^+ - \lambda_- Z_1^- - \frac{\lambda_+^2}{2} Z_2^{++} - \frac{\lambda_-^2}{2} Z_2^{--} - \lambda_+ \lambda_- Z_2^{+-} + O(\lambda^3), \quad (25)$$

where the coefficients are given by

$$Z_1^+ = \frac{1}{V} \int d\mathbf{r} \langle Q_+(\mathbf{r}) \rangle \quad (26)$$

and an analogous formula for  $Z_1^-$ , and

$$Z_2^{++} = \frac{1}{V} \int d\mathbf{r} d\mathbf{r}' \left\{ \langle Q_+(\mathbf{r}) Q_+(\mathbf{r}') \rangle - \langle Q_+(\mathbf{r}) \rangle \langle Q_+(\mathbf{r}') \rangle \right\} \quad (27)$$

and similar formulas for  $Z_2^{--}$  and  $Z_2^{+-}$ . The symbol  $O(\lambda^3)$  in equation (25) means that any other contribution to the expansion will be of order 3 or higher, *i.e.*, with terms like  $\lambda_+^3$ ,  $\lambda_+^2 \lambda_-$ , etc. Clearly, the expectation values in equations (26, 27) contain additional dependencies on the fugacity  $\lambda_\alpha$  *via* the DH propagator, equation (16), but, and

this stands at the very core of our method, all expectation values can itself be expanded with respect to  $\lambda_\alpha$  and have finite values as  $\lambda_\alpha \rightarrow 0$ .

In order to do a full expansion of equation (25), one needs first to calculate the coefficients  $Z_1^+$ , etc., in equation (25), for which the averages given in the appendix are needed, *cf.* equations (78–82). Since  $v_c(0) - v_{\text{DH}}(0) = \Delta v_0$ , we obtain

$$Z_1^\alpha = e^{q_\alpha^2 \Delta v_0 / 2} - 1 - \frac{1}{2} q_\alpha^2 \Delta v_0 \quad (28)$$

and

$$Z_2^{\alpha\beta} = \int d\mathbf{r} \left\{ e^{[q_\alpha^2 + q_\beta^2] \Delta v_0 / 2} \left[ e^{-q_\alpha q_\beta [\omega_{\alpha\beta}(\mathbf{r}) + \alpha\beta v_{\text{DH}}(\mathbf{r})]} - 1 \right] + \frac{1}{2} q_\alpha^2 q_\beta^2 v_{\text{DH}}^2(\mathbf{r}) \left[ 1 - e^{q_\alpha^2 \Delta v_0 / 2} - e^{q_\beta^2 \Delta v_0 / 2} \right] \right\}. \quad (29)$$

Note that

$$v_{\text{DH}}(\mathbf{r}) = \frac{\ell_B}{r} e^{-\Delta v_0 r / \ell_B} \quad (30)$$

was defined (through its inverse function) in equation (16). We now introduce the hard-core through the short-range potentials

$$\omega_{\alpha\beta}(\mathbf{r}) = \begin{cases} +\infty & \text{if } r < (d_\alpha + d_\beta)/2, \\ 0 & \text{otherwise,} \end{cases} \quad (31)$$

where the indices  $\alpha$  and  $\beta$  stand again for  $+$  and  $-$ ;  $d_+$  and  $d_-$  are respectively the (effective) ionic diameters of the positive and negative particles.

Since the expressions for  $Z_1^+$ , etc., do depend on  $\lambda_+$  and  $\lambda_-$ , and in order to have a consistent expansion of  $g$  in the fugacities, one has to expand equations (28, 29) in powers of  $\lambda_+$  and  $\lambda_-$ , up to the appropriate order, before inserting them into  $g$ , equation (25). By doing this consistently up to order 5/2 in the fugacities, one obtains the rescaled grand-canonical free energy density

$$\begin{aligned} \tilde{g} \equiv d_+^3 g = & -\tilde{\lambda}_+ - \tilde{\lambda}_- - m_1 \tilde{\lambda}_+^2 - m_2 \tilde{\lambda}_-^2 - m_3 \tilde{\lambda}_+ \tilde{\lambda}_- - \left[ n_1 \tilde{\lambda}_+^2 \right. \\ & + n_2 \tilde{\lambda}_-^2 + n_3 \tilde{\lambda}_+ \tilde{\lambda}_- \left. \right] \Delta v_0 - \left[ p_1 \tilde{\lambda}_+^2 + p_2 \tilde{\lambda}_-^2 + p_3 \tilde{\lambda}_+ \tilde{\lambda}_- \right] \ln \Delta v_0 \\ & - \left[ r_1 \tilde{\lambda}_+^2 + r_2 \tilde{\lambda}_-^2 + r_3 \tilde{\lambda}_+ \tilde{\lambda}_- \right] \Delta v_0 \ln \Delta v_0 - \left[ s_1 \tilde{\lambda}_+ \right. \\ & + s_2 \tilde{\lambda}_- \left. \right] \Delta v_0^2 - \left[ t_0 + t_1 \tilde{\lambda}_+ + t_2 \tilde{\lambda}_- \right] \Delta v_0^3 + \Omega_0 \left[ q_+ \tilde{\lambda}_+ - q_- \tilde{\lambda}_- \right] \\ & - \Delta v_0 \left\{ \Omega_1 \left[ q_+ \tilde{\lambda}_+ - q_- \tilde{\lambda}_- \right]^2 - \Omega_2 \left[ q_+^4 \tilde{\lambda}_+^2 + q_-^4 \tilde{\lambda}_-^2 \right. \right. \\ & \left. \left. - 2q_+ q_- \left[ \frac{q_+^2 + q_-^2}{2} \right] \tilde{\lambda}_+ \tilde{\lambda}_- \right] \right\} + O(\tilde{\lambda}^3), \quad (32) \end{aligned}$$

where we used dimensionless fugacities

$$\tilde{\lambda}_\pm \equiv d_\pm^3 \lambda_\pm \quad (33)$$

and

$$\tilde{\lambda}_- \equiv d_+^3 \lambda_- . \quad (34)$$

In this expansion, we utilized that  $\Delta v_0$  scales like  $\tilde{\lambda}_-^{1/2}$ . The coefficients  $m_1$ , etc. are given explicitly in the appendix, equations (83–92).

The coefficients  $\Omega_0$  and  $\Omega_1$  are given by the divergent integrals

$$\begin{aligned} \Omega_0 &= 2\pi \ell_B \int_0^\infty dr r \\ \Omega_1 &= 2\pi \int_0^\infty dr r^2 . \end{aligned} \quad (35)$$

These terms are present in equation (32) because global charge neutrality has not been yet demanded. By imposing this condition, these divergent terms cancel exactly, as is shown next.

## 2.2 Imposing electroneutrality

The electroneutrality condition ensures that the global charge of the system is zero, *i.e.*,  $q_+ N_+ = q_- N_-$ . In the grand-canonical ensemble,  $N_+$  and  $N_-$  are no longer fixed numbers but average values. This means that the electroneutrality condition in the grand-canonical ensemble is given by

$$q_+ \langle N_+ \rangle = q_- \langle N_- \rangle . \quad (36)$$

Defining the rescaled ion density

$$\tilde{c}_+ = d_+^3 \langle N_+ \rangle / V , \quad (37)$$

it is easy to show that

$$\tilde{c}_+ = -\tilde{\lambda}_+ \frac{\partial \tilde{g}}{\partial \tilde{\lambda}_+} \quad (38)$$

with an analogous formula for  $\tilde{c}_-$ ;  $\tilde{c}_+$  is the volume fractions of the positive ions [12]. As one imposes equation (36), the fugacities will depend on each other in a non-trivial way such that the system is, on average, neutral. If the system were totally symmetric (*i.e.*,  $q_+ = q_-$  and  $d_+ = d_-$ ), this dependence would be given by the relation  $\tilde{\lambda}_+ = \tilde{\lambda}_-$  [10]. However, this is not the case here: one needs to find the relation between the fugacities order-by-order.

First, assume that  $\tilde{\lambda}_-$  can be expanded in terms of  $\tilde{\lambda}_+$  such that

$$\begin{aligned} \tilde{\lambda}_- &= a_0 \tilde{\lambda}_+ + a_1 \tilde{\lambda}_+^{3/2} + a_2 \tilde{\lambda}_+^2 + a_3 \tilde{\lambda}_+^2 \ln \tilde{\lambda}_+ \\ &+ a_4 \tilde{\lambda}_+^{5/2} + a_5 \tilde{\lambda}_+^{5/2} \ln \tilde{\lambda}_+ + O(\tilde{\lambda}_+^3) . \end{aligned} \quad (39)$$

This is inspired by the expanded form of the grand-canonical free energy  $\tilde{g}$ , equation (32).

After calculating  $\tilde{c}_+$  and  $\tilde{c}_-$  from  $\tilde{g}$ , equation (32), according to equation (38), and insertion into the electroneutrality condition (Eq. (36)), we substitute the fugacity  $\tilde{\lambda}_-$  by its expanded form equation (39). This leads to the expanded form (up to order  $\tilde{\lambda}_+^{5/2} \ln \tilde{\lambda}_+$ ) of the electroneutrality condition. Solving it consistently, order-by-order, yields the values of the coefficients in equation (39), which ensure electroneutrality order by order. For instance, at linear order in  $\tilde{\lambda}_+$ , the expanded form of equation (36) reads

$$q_+ - a_0 q_- = 0 , \quad (40)$$

which naturally gives  $a_0 = q_+ / q_-$ . With the knowledge of  $a_0$ , one can solve the next-order term (in this case  $\tilde{\lambda}_+^{3/2}$ ) and obtain the value of  $a_1$ , and so on. The resulting coefficients  $a_0$  up to  $a_5$  are given in Appendix — *cf.* equations (93–98).

As this order-by-order neutrality condition is imposed, one notices that the terms  $\Omega_0$  and  $\Omega_1$  in equation (32) are exactly canceled in a natural way, without any further assumptions. The resulting expression for  $\tilde{g}$ , now expanded only in one of the fugacities (in this case  $\tilde{\lambda}_+$ ), is then a well behaved expansion (we omit this expression here since it is quite lengthy). With this, one can finally obtain the canonical free energy (as a density expansion) through a Legendre transform.

## 2.3 The canonical free energy

In order to obtain the free energy in the canonical ensemble, we use the back-Legendre-transformation

$$\tilde{f} = \tilde{g} + \tilde{c}_+ \ln(\tilde{\lambda}_+) + \tilde{c}_- \ln(\tilde{\lambda}_-) , \quad (41)$$

where the dimensionless canonical free energy density is

$$\tilde{f} \equiv d_+^3 F / V k_B T \quad (42)$$

( $F$  is the canonical free energy). Note that at this point  $\tilde{\lambda}_-$  is a function of  $\tilde{\lambda}_+$ , according to equation (39).

The first step to obtain  $\tilde{f}$  is to invert the expression given in equation (38) such that  $\tilde{\lambda}_+$  is obtained as an expansion in  $\tilde{c}_+$ , neglecting any terms of order  $\tilde{c}_+^3$  or higher. With this, we obtain  $\tilde{\lambda}_+ = \tilde{\lambda}_+(\tilde{c}_+)$ : plugging this into equation (41), we finally obtain  $\tilde{f}$ , which reads

$$\begin{aligned} \tilde{f} &= \tilde{f}_{\text{id}} + B_{\text{DH}} \tilde{c}_+^{3/2} + B_2 \tilde{c}_+^2 + B_{2\log} \tilde{c}_+^2 \ln \tilde{c}_+ \\ &+ B_{5/2} \tilde{c}_+^{5/2} + B_{5/2\log} \tilde{c}_+^{5/2} \ln \tilde{c}_+ + O(\tilde{c}_+^3) . \end{aligned} \quad (43)$$

Defining the valence ratio parameter

$$\eta \equiv q_- / q_+ , \quad (44)$$

the diameter ratio parameter

$$\xi \equiv d_- / d_+ , \quad (45)$$

and the coupling parameter

$$\epsilon_+ \equiv q_+^2 \ell_B / d_+ \quad (46)$$

(which is the ratio between the Coulomb energy at contact between two positive ions and the thermal energy  $k_B T$ ), the coefficients in equation (43) can be explicitly written as

$$\tilde{f}_{\text{id}} = \tilde{c}_+ \ln \tilde{c}_+ + \frac{\tilde{c}_+}{\eta} \ln \left( \frac{\tilde{c}_+}{\eta} \right) - \left[ 1 + \frac{1}{\eta} \right] \tilde{c}_+ \quad (47)$$

which is the ideal contribution to the free energy,

$$B_{\text{DH}} = -\frac{2}{3} \sqrt{\pi \epsilon_+^3 [1 + \eta]^3} \quad (48)$$

which is the coefficient of the Debye-Hückel limiting law term (order 3/2 in  $\tilde{c}_+$ ). It is useful to define the function

$$H(x) = \frac{11}{6} - 2\gamma + \frac{1}{x^3} e^{-x} [2 - x + x^2] - \Gamma(0, x) - \ln x, \quad (49)$$

where  $\gamma$  is the Euler's constant and  $\Gamma(a, b)$  is the incomplete Gamma-function [13]. Using  $H(x)$ , the higher order coefficients can be explicitly written as

$$B_2 = -\frac{\pi}{3} \epsilon_+^3 \left\{ -H(\epsilon_+) - \ln \epsilon_+ + 2\eta^2 \left[ H \left( -\frac{2\eta\epsilon_+}{1+\xi} \right) + \ln \left( \frac{2\eta\epsilon_+}{1+\xi} \right) \right] - \eta^4 \left[ H \left( \frac{\eta^2\epsilon_+}{\xi} \right) + \ln \left( \frac{\eta^2\epsilon_+}{\xi} \right) \right] - 2\eta^2 [1 - \eta^2] \ln \eta + \frac{1}{2} [1 - \eta^2]^2 \ln(36\pi\epsilon_+^3 [1 + \eta]) \right\} \quad (50)$$

$$B_{2\log} = -\frac{\pi}{6} \epsilon_+^3 [1 - \eta^2]^2, \quad (51)$$

$$B_{5/2} = \frac{2}{3} [\pi\epsilon_+^3]^{3/2} [1 + \eta]^{1/2} \left\{ \frac{5}{8} + H(\epsilon_+) + \ln(\epsilon_+) + \eta^6 \left[ \frac{5}{8} + H \left( \frac{\eta^2\epsilon_+}{\xi} \right) + \ln \left( \frac{\eta^2\epsilon_+}{\xi} \right) \right] + 2\eta^3 \left[ \frac{5}{8} + H \left( -\frac{2\eta\epsilon_+}{1+\xi} \right) + \ln \left( \frac{2\eta\epsilon_+}{1+\xi} \right) \right] + \frac{1}{8} [1 + \eta]^2 [5 - 12\eta + 17\eta^2 - 12\eta^3 + 5\eta^4] - 2\eta^3 [1 + \eta^3] \ln \eta - \frac{1}{2} [1 + \eta^3]^2 \ln(64\pi\epsilon_+^3 [1 + \eta]) \right\} \quad (52)$$

and

$$B_{5/2\log} = -\frac{1}{3} [\pi\epsilon_+^3]^{3/2} [1 + \eta]^{1/2} [1 + \eta^3]^2. \quad (53)$$

The free energy equation (43) is the exact low density expansion of the asymmetric TCPHC and constitutes the

main result of this paper. The only parameter that is demanded to be small is the ion density  $\tilde{c}_+$ , that means, this result is non-perturbative in the coupling  $\epsilon_+$ , charge ratio  $\eta$  and size ratio  $\xi$ .

We chose the positive ions as the ‘‘reference species’’ (*i.e.*, the expansion is done with respect to  $\tilde{c}_+$ ) without any loss of generality, since the relation between  $\tilde{c}_+$  and  $\tilde{c}_-$  is fixed through the electroneutrality condition. As consistency checks, we notice that our expression for  $\tilde{f}$  is symmetric, as expected, with respect to the simultaneous exchange of  $d_+$  with  $d_-$  and  $q_+$  with  $q_-$ . Also, in the limit  $d_+ = d_-$  and  $q_+ = q_-$ , we recover the same expression as previously calculated in reference [10] for totally symmetric systems. Finally, as one turns off the charges in the system (or equivalently, as one takes the limit  $\epsilon_+ \rightarrow 0$ ), the pure hard-core fluid is recovered, *i.e.*,  $\tilde{f}$  becomes the usual virial expansion with  $B_{3/2}$  and  $B_{5/2}$  equal to zero and  $B_2$  given by the second virial coefficient of a two-component hard-core gas,  $B_2^{\text{HC}} = B_2(\epsilon_+ \rightarrow 0)$ , which in our units reads

$$B_2^{\text{HC}} = \frac{\pi}{3} \left[ 2 + \frac{2\xi^3}{\eta^2} + \frac{[1 + \xi]^3}{2\eta} \right]. \quad (54)$$

This limit can be also understood as the high-temperature regime: as the thermal energy largely exceeds the Coulomb energy at contact, the hard core interaction becomes the only relevant interaction between the particles.

## 3 Results

### 3.1 The virial coefficients

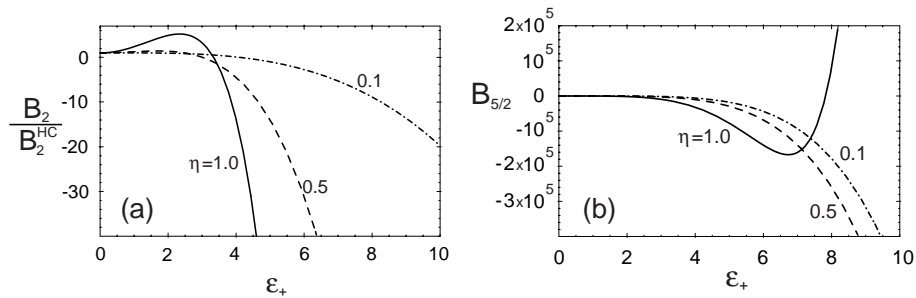
The behavior of the coefficients  $B_2$  and  $B_{5/2}$  as functions of the coupling parameter  $\epsilon_+$  are depicted in Figures 1 and 2, the behavior of  $B_{2\log}$  and  $B_{5/2\log}$  is rather trivial and not shown graphically. In Figure 1 the ionic diameters of positive and negative ions are kept equal,  $d_+ = d_-$ , and the ratio between the charge valences,  $\eta$ , is varied, while in Figure 2 the charge valences are equal and the ratio between the ionic diameters is varied. These figures highlight the fact that both coefficients diverge as  $\epsilon_+$  goes to infinity. In this limit,

$$B_2 \approx -\frac{\pi}{4} \frac{[1 + \xi]^4}{\eta^2} \frac{1}{\epsilon_+} \exp \left( \frac{2\eta\epsilon_+}{1 + \xi} \right) \quad (55)$$

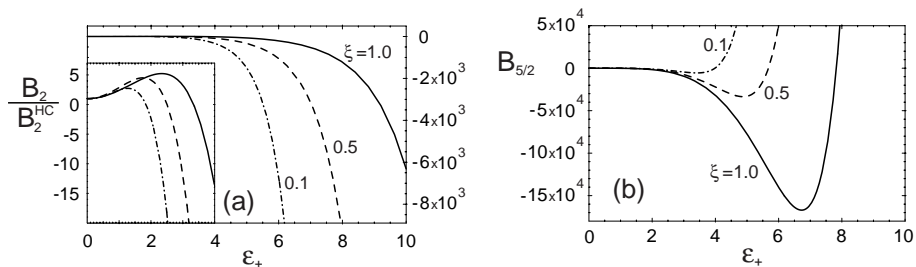
and

$$B_{5/2} \approx \frac{\pi^{3/2}}{2} \frac{[1 + \xi]^4 \sqrt{1 + \eta}}{\eta} \sqrt{\epsilon_+} \exp \left( \frac{2\eta\epsilon_+}{1 + \xi} \right). \quad (56)$$

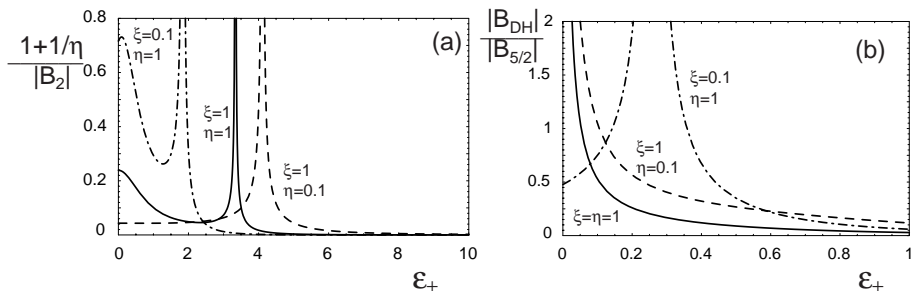
Note that  $B_{5/2}$  diverges faster (and with opposite sign) than  $B_2$ , reflecting that this low-density expansion is badly converging, as we will discuss further below. The exponentially divergent behavior of  $B_2$  and  $B_{5/2}$  when  $\epsilon_+ \rightarrow \infty$  is due to the increasing importance of the interaction between oppositely charged particles (ionic pairing) as the coupling parameter increases [14–16], corresponding for instance to lower temperatures. This is confirmed



**Fig. 1.** Coefficients (a)  $B_2$  and (b)  $B_{5/2}$  of the free energy expansion equation (43) as functions of  $\epsilon_+$  for different values of  $\eta \equiv q_-/q_+$  (with  $d_+ = d_-$  or  $\xi = 1$ ). Notice that  $B_2$  is normalized by the second virial coefficient  $B_2^{\text{HC}}$  of a pure two-component hard-core gas.



**Fig. 2.** Coefficients (a)  $B_2$  and (b)  $B_{5/2}$  of the free energy equation (43) as functions of  $\epsilon_+$  for different values of  $\xi \equiv d_-/d_+$  (with  $q_+ = q_-$ ). Notice that  $B_2$  is normalized by the second virial coefficient  $B_2^{\text{HC}}$  of a pure two-component hard-core gas. The inset in (a) shows the behavior of  $B_2$  close to  $\epsilon_+ = 0$  on a different scale.



**Fig. 3.** Ratios of the coefficients in the low-density expansion, which serve as rough estimates of the maximal rescaled ion density  $\tilde{c}_+$  up to which the expansion is valid (see text).

by noting that the argument in the exponential occurring in both asymptotic forms equations (55, 56), *viz.*  $2\eta\epsilon_+[1+\xi]$ , can be re-expressed as  $2q_+q_- \ell_B/[d_+ + d_-]$ , which is the coupling between positive and negative ions (in this case, the ratio between the Coulomb contact energy between oppositely charged ions and the thermal energy  $k_B T$ ).

To estimate roughly the ionic density  $\tilde{c}_+$  up to which the expansion in equation (43) is expected to be valid, we use the following simple criterion: the terms proportional to  $B_{\text{DH}}$  and  $B_{5/2}$  give the same contribution to the free energy when  $|B_{\text{DH}}|\tilde{c}_+^{3/2} = |B_{5/2}|\tilde{c}_+^{5/2}$  or, in other words, at a critical density  $\tilde{c}_+ = |B_{\text{DH}}/B_{5/2}|$ . Note that we set up this criterion separately for the integer and fractional coefficients, since the scaling behavior for the two different classes is very different and no meaningful results can be obtained by mixing them. The analogous criterion for the

integer terms leads to a critical density  $\tilde{c}_+ = (1+1/\eta)/|B_2|$  where for the ideal contribution in equation (47) we replace the logarithmic term by a linear one (which is the ideal contribution to the osmotic pressure). We tacitly assume that the higher-order terms (which we have not calculated) show the same relative behavior, an assumption which seems plausible to us but which we cannot check. In Figure 3 we show the estimate for the critical densities obtained from (a) the ratio of the integer terms and from (b) the ratio of the fractional terms. In (a) the critical density settles at a finite value for vanishing coupling parameter  $\epsilon_+ \rightarrow 0$ , and decreases to zero for increasing coupling parameter (the divergence in the curves at finite value of  $\epsilon_+$  is not significant since it is caused by a change in sign of  $B_2$ ). This means that for small coupling constants the integer coefficients are expected to show regular convergence behavior for not too large concentrations. For large values of  $\epsilon_+$  on the other hand, the range of densities which can

be described by the expansion quickly goes to zero. In (b) the displayed behavior is more complex. The expansion of the coefficient  $B_{5/2}$  for small values of the coupling parameter  $\varepsilon_+$  leads to

$$B_{5/2} = \pi^{3/2} \sqrt{1+\eta} (\xi-1)^2 (\xi+1) \varepsilon_+^{3/2} - \pi^{3/2} \sqrt{1+\eta} (2+2\eta^2\xi^2 + \eta(1+\xi)^2) \varepsilon_+^{5/2} \quad (57)$$

plus terms with scale as higher powers in  $\varepsilon_+$ . The important point is that except for the singular case  $\xi = 1$ , that is for particles of identical radius, the leading term of  $B_{5/2}$  scales as  $\varepsilon_+^{3/2}$  and thus identical to the lower leading term  $B_{\text{DH}}$ . In the case  $\xi = 1$  the leading term scales as  $\varepsilon_+^{5/2}$  and thus shows a different scaling behavior than  $B_{\text{DH}}$ . In Figure 3b this difference is clearly seen: for  $\xi = 1$  the curves diverge as  $\varepsilon_+ \rightarrow 0$  and the convergence of the fractional density terms in the series is expected to be guaranteed. For  $\xi = 0.1$  the behavior is similar to the results in (a), showing a saturation at a finite value as  $\varepsilon_+ \rightarrow 0$  (and an insignificant divergence at finite  $\varepsilon_+$ ). All curves go to zero as the coupling parameter grows. In conclusion, the reliability of the low-density expansion becomes worse as the coupling parameter increases, but one can always find a window of small densities within which the expansion should work.

Finally, we mention that in a different calculation scheme, based on an integral-equation-procedure within the so-called mean-spherical-model (MSM) approximation, similar results for the free energy and other thermodynamic functions have been obtained [17–19]. Those results also reproduce the limiting laws, namely the hard-core behavior as the coupling parameter goes to zero, and the leading Debye-Hückel correction at low density, and give a quite satisfactory description even for much larger values of the density. It is important to note, however, that the next-leading (beyond Debye-Hückel) terms in the low-density expansion are not correctly reproduced by the MSM approximation.

### 3.2 The colloidal limit

In colloidal suspensions, flocculation or coagulation (driven by attractive van der Waals interaction between colloidal particles) can be prevented by the presence of repulsive electrostatic forces. These suspensions are typically quite dilute, with volume fractions of colloidal particles usually not higher than a few percent. The macro-particles normally have dimensions [20–22] ranging from 10 to 1 000 nm and charges of several thousands  $e$  (elementary charge), with much smaller counterions that have a charge of a few  $e$ . In such systems the charge and size asymmetry between ions and counterions is immense; the TCPHC with unconstrained charge and size asymmetry is a suitable model for dilute colloidal suspensions, and the free energy equation (43) can be used to study the thermodynamic behavior of such systems.

With this in mind, let us take the following limit: assume the parameters describing the positive ions (representing the colloids)  $d_+$ ,  $q_+$  and  $\tilde{c}_+$  fixed, and make both

$\eta \equiv q_-/q_+$  and  $\xi \equiv d_-/d_+$  vanishingly small (this has to be done with some care, since the limit  $\eta = 0$  is not well-defined). One can then rewrite the free energy equation (43) up to second order in  $\tilde{c}_+$  as

$$\tilde{f}^{\text{cl}} = \tilde{f}_{\text{id}} + B_{\text{DH}}^{\text{cl}} \tilde{c}_+^{3/2} + (B_2^{\text{HC,cl}} + B_2^{\text{cl}}) \tilde{c}_+^2 + B_{2\log}^{\text{cl}} \tilde{c}_+^2 \log \tilde{c}_+ + O(\tilde{c}_+^{5/2}) \quad (58)$$

where  $\tilde{f}_{\text{id}}$  is the ideal term equation (47) and

$$B_2^{\text{HC,cl}} = \frac{\pi}{3} \left[ \frac{2\xi^3}{\eta^2} + \frac{[1+\xi]^3}{2\eta} \right] \quad (59)$$

is the hard core contribution due to the counterion-counterion and macroion-counterion interactions only, without the macroion-macroion contributions, which explains the difference to the full hard-core virial coefficient in equation (54). It is necessary to treat this term separately, as it does not have a well-defined behavior in the double limit  $\eta \rightarrow 0$  and  $\xi \rightarrow 0$ . The other coefficients in equation (58) follow from equations (48, 50, 51) by performing the limits  $B_{\text{DH}}^{\text{cl}} = B_{\text{DH}}(\eta \rightarrow 0, \xi \rightarrow 0)$ ,  $B_2^{\text{cl}} = B_2(\eta \rightarrow 0, \xi \rightarrow 0) - B_2^{\text{HC,cl}}$ ,  $B_{2\log}^{\text{cl}} = B_{2\log}(\eta \rightarrow 0, \xi \rightarrow 0)$ , and are given by

$$B_{\text{DH}}^{\text{cl}} = -\frac{2}{3} \sqrt{\pi \epsilon_+^3}, \quad (60)$$

$$B_2^{\text{cl}} = \frac{\pi}{3} \epsilon_+^3 \left\{ H(\epsilon_+) - \frac{1}{2} \ln(36\pi\epsilon_+) \right\} + \frac{\pi}{2} \epsilon_+, \quad (61)$$

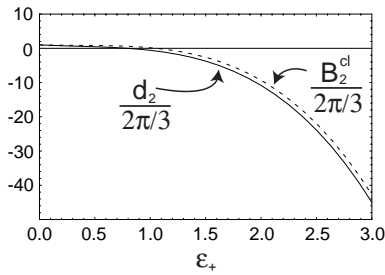
$$B_{2\log}^{\text{cl}} = -\frac{\pi}{6} \epsilon_+^3. \quad (62)$$

It is to be noted that the contributions in  $\tilde{f}_{\text{id}}$  and  $B_2^{\text{HC,cl}}$  contain terms that scale with  $1/\eta$ . For dilute colloidal systems therefore the ideal contribution of the counterions dominates the free energy and cannot be neglected. Effects due to the electrostatic interaction between the ions are corrections to the ideal behavior.

At this point we briefly introduce yet another model that is also widely used to describe charged systems. It is the one-component plasma (OCP), which in its simplest form consists of a collection of  $N$  equally charged point-like particles immersed in a neutralizing background that ensures the global charge neutrality of the system (in the TCPHC electroneutrality is ensured by oppositely charged particles). The OCP, or its quantum mechanical counter-part (“jellium”) has been used in different contexts in physics, as for instance to describe degenerate stellar matter (the interior of white dwarfs or the outer layer of neutron stars) and the interior of massive planets like Jupiter. Another example comes from condensed matter physics, where jellium is often used as a reference state when calculating the electronic structure of solids. For reviews see references [23–25].

When the particles have a hard core, the OCP is called one-component hard-core plasma (OCPHC): what we will see next is that the electrostatic contribution to the free





**Fig. 4.** Coefficients  $B_2^{\text{cl}}$  (Eq. (65)) and  $d_2$  (pure OCPHC from Ref. [10]) as functions of the coupling  $\epsilon_+$ . Notice that both coefficients are normalized by the second virial of a pure one-component hard-core gas.

energy in dilute colloidal suspensions can be almost described through the OCPHC. If we compare the coefficients equations (51–60) with the ones previously obtained [10] for the low density expansion of the OCPHC

$$\tilde{f}_{\text{OCPHC}} = \tilde{c} \ln \tilde{c} - \tilde{c} + d_{3/2} \tilde{c}^{3/2} + d_2 \tilde{c}^2 + d_{\ln 2} \tilde{c}^2 \ln \tilde{c} + O(\tilde{c}^{5/2}), \quad (63)$$

where  $\tilde{c}$  is the volume fraction of particles in the OCPHC, we find (*cf.* Eqs. (51, 56, 57) of Ref. [10])

$$B_{\text{DH}}^{\text{cl}} = d_{3/2}, \quad (64)$$

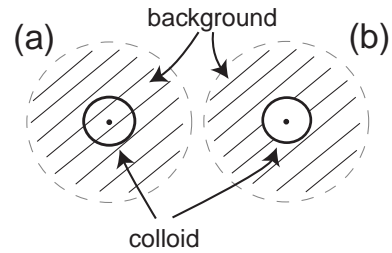
$$B_2^{\text{cl}} = d_2 + \frac{\pi}{2} \epsilon_+ \quad (65)$$

and

$$B_{2\log}^{\text{cl}} = d_{\ln 2}, \quad (66)$$

where the notation used in reference [10] is kept in equation (63) and on the rhs of equations (64–66). The comparison between  $B_2^{\text{cl}}$  and  $d_2$  is shown in Figure 4, where we rescaled both coefficients by their value at vanishing coupling,  $d_2(\epsilon_+ = 0) = B_2^{\text{cl}}(\epsilon_+ = 0) = 2\pi/3$ . This shows that, including the order  $\tilde{c}_+^2 \ln \tilde{c}_+$  (*i.e.*, for very dilute colloidal suspension), the OCPHC is almost recovered, except an additional term in equation (65). This additional term can be quite important at intermediate values of  $\epsilon_+$  since it changes the sign of the virial coefficient, as seen in Figure 3.

This is in fact what one would intuitively expect: each macroion has around it a very large number of small neutralizing counterions which act like a background. Our results show that the background formed by counterions, which is a deformable background since the counterion distribution is not uniform, acts almost like a rigid homogeneous background, as it is assumed in OCP calculations. However, there is a small difference between the TCPHC in this limit and the OCPHC: in the latter, the background penetrates the particles, while in the TCPHC it cannot (for a thorough discussion of this difference see [26]). In our calculation, this is reflected in the  $\tilde{c}_+^2$  term, where  $\pi\epsilon_+/2$  is the positive extra cost in the free energy that the OCPHC has to pay (at this order) to expel the background from the hard-core particles. The first consequence of the



**Fig. 5.** A colloidal particle (with diameter  $d_+$  and charge valence  $q_+$ ) in the OCPHC model (a) with and (b) without the penetrating background.

non-penetrating background is that the effective charge of the colloids increases by exactly the amount of background charge that is expelled from the colloidal interior. A simple calculation shows that the increased effective charge  $q_+^{\text{eff}}$  of colloids for the non-penetrating background turns out to be

$$q_+^{\text{eff}} = \frac{q_+}{1 - \pi\tilde{c}_+/6}. \quad (67)$$

In the low-density expansion of the OCPHC, equation (63), one would have to reexpand all coefficients with respect to the density, using equation (67). However, since the leading term which depends on the charge  $q_+$  goes like  $\tilde{c}_+^{3/2}$ , the effect would come in at order  $\tilde{c}_+^{5/2}$  and therefore is not responsible for the additional term in equation (65). The reason for the extra term in equation (65) has to do with an increase of the self-free-energy of a colloidal particle, which can be understood in the following way: assume that the OCPHC is very dilute, such that each colloidal particle and its neutralizing background form a neutral entity (in the spirit of the cell model, see for instance Ref. [27]) that can be regarded independent from the other particles. The free energy difference *per colloidal particle* between a system without penetrating background and with penetrating background (Fig. 5) is then given by two contributions: one coming from the entropy lost by the background (formed by  $N_-$  counterions) since it cannot penetrate the macroions, given by

$$\frac{\Delta F_{\text{en}}}{N_+ k_{\text{B}} T} = -\frac{N_-}{N_+} \ln(1 - \pi\tilde{c}_+/6) = \frac{N_-}{N_+} \frac{\pi\tilde{c}_+}{6} + O(\tilde{c}_+^2). \quad (68)$$

This is one of the contributions to the second virial coefficient due to the hard-core interaction, and already included in equation (59). This term does not depend on the coupling  $\epsilon_+$ , and therefore has nothing to do with the extra term in equation (65). The second contribution to the self-energy is electrostatic in origin. Defining the charge distributions for the situations in Figure 4 as  $\rho_a(\mathbf{r}) = q_+ \delta(\mathbf{r}) - \rho_0$  when the background can enter the colloids and  $\rho_b(\mathbf{r}) = q_+ \delta(\mathbf{r}) - \rho_0 [1 - \theta(d_+/2 - r)]$  when the background cannot enter the colloids (where  $\rho_0 = c_+ q_+$  is the background charge density and  $\theta(x) = 1$  if  $x > 0$  and 0 otherwise), the electrostatic self-energy difference

between the two cases reads

$$\begin{aligned} \frac{\Delta F_{\text{el}}}{N_+ k_B T} &= \frac{1}{2} \int d\mathbf{r} d\mathbf{r}' \left\{ \rho_b(\mathbf{r}) v_c(\mathbf{r} - \mathbf{r}') \rho_b(\mathbf{r}') \right. \\ &\quad \left. - \rho_a(\mathbf{r}) v_c(\mathbf{r} - \mathbf{r}') \rho_a(\mathbf{r}') \right\} \\ &= \rho_0 q_+ 4\pi \ell_B \int_0^{d_+/2} dr r + O(c_+^2) \\ &= \frac{\pi}{2} \tilde{c}_+ \epsilon_+ + O(c_+^2). \end{aligned} \quad (69)$$

It follows that the free energy difference per volume is given by

$$\Delta \tilde{f} = \frac{\tilde{c}_+ \Delta F_{\text{el}}}{N_+ k_B T} = \frac{\pi \tilde{c}_+^2 \epsilon_+}{2}. \quad (70)$$

Note that this is a positive contribution to the  $\tilde{c}^2$  term with the same coefficient as the extra term in equation (65). Therefore, we conclude that this extra term is due to the electrostatic energy associated with expelling the background from the colloidal volume. This is in accord with more general results on the difference between OCPHC models with penetrating and excluded neutralising backgrounds [26]. In summary, whenever using the OCPHC to describe highly asymmetric charged systems such as colloids, one has to take into account the exclusion of the background from the macroions. As we demonstrated, if this is taken into account, then the TCPHC maps exactly (at least up to order  $\tilde{c}_+^2$ ) onto the OCPHC.

### 3.3 Ionic activity and diameters

In electrochemistry, it is usual to define the mean activity  $\lambda_{\pm}$  of an electrolyte as

$$\lambda_{\pm} \equiv \left[ \lambda_+^{q_-} \lambda_-^{q_+} \right]^{1/(q_+ + q_-)}, \quad (71)$$

where  $\lambda_+$  and  $\lambda_-$  are respectively the fugacities of the positive and the negative ions. The mean activity coefficient  $f_{\pm}$  is the ratio between the mean activity of the electrolyte and that of an ideal gas, in general given by

$$f_{\pm} = \exp \left( \frac{q_-}{q_+ + q_-} \frac{\partial \tilde{f}_{\text{ex}}}{\partial \tilde{c}_+} \right), \quad (72)$$

for a two-component system where the positively charged particles are used as reference species in the same way as in equation (43).  $\tilde{f}_{\text{ex}}$ , the excess free energy, is the difference between the free energy of the interacting systems and the free energy of an ideal gas, *i.e.*,  $\tilde{f}_{\text{ex}} = \tilde{f} - \tilde{f}_{\text{id}}$ .

There are different ways of measuring  $f_{\pm}$ , as for instance, through the change of the freezing point of the solvent (usually water) with the addition of salt [28], by measuring the change of the potential difference between the electrodes of a concentration cell as salt is added [28,29] (potentiometry), or by direct measurement of the solvent

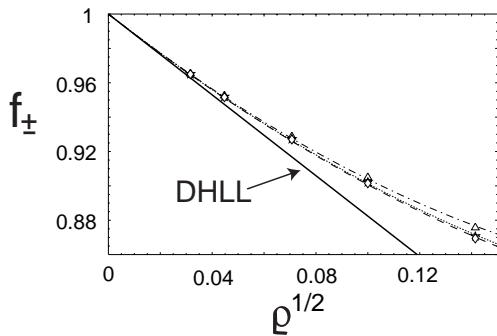
activity through vapor exchange between a solution with known activity and the sample [28,30] (isopiestic). Although dating from the early nineteen hundreds, these are still the most common techniques used today, especially the potentiometry, which is regarded as the most precise technique of all. The values of  $f_{\pm}$  are tabled as function of the salt concentration for many different electrolytes [28,31,32].

From the free energy equation (43) and the definition equation (72) we can obtain the low density expansion for the mean activity coefficient of a  $q_+ : q_-$  salt. To compare with experimental results, it is useful to note that  $\tilde{c}_+ = 6.022 \times 10^{-4} q_- d_+^3 \varrho$ , where  $d_+$  is in angströms and  $\varrho$  is the salt concentration in moles/liter. After the appropriate expansion,  $f_{\pm}$  reads

$$\begin{aligned} f_{\pm} &= 1 + \nu_{\text{DH}} \varrho^{1/2} + \nu_1 \varrho + \nu_{1\log} \varrho \ln \varrho + \nu_{3/2} \varrho^{3/2} \\ &\quad + \nu_{3/2\log} \varrho^{3/2} \ln \varrho + O(\varrho^2) \end{aligned} \quad (73)$$

(the order 3/2 in the mean activity coefficient is the one consistent with a free energy given up to 5/2). The coefficients  $\nu_{\text{DH}}$ ,  $\nu_1$ , etc. are given in Appendix, equations (99–103). Experimentally determined activity coefficients are typically measured at constant pressure, while the theoretical results are obtained for constant volume. Note that in the limit of vanishing density, the distinction between the activities calculated in the Lewis-Randall or in the McMillan-Mayer descriptions (constant pressure versus constant volume ensemble) becomes negligible [33] in comparison to the first corrections to ideal behavior. The experimental data at low density can be directly compared with the theory without the need to convert between the two ensembles.

At infinite dilution, the mean activity coefficient equation (73) goes to 1, which is the prediction for an ideal gas. The first correction to the ideal behavior is the term  $\nu_{\text{DH}} \varrho^{1/2}$ , which is the prediction one obtains from the Debye-Hückel limiting law (DHLL), and is independent of the ionic diameters, see equation (99). This means that there is always a range of concentrations where different salts (but with the same  $q_+$  and  $q_-$ ) will deviate from ideality, but have the same activity. At higher concentrations, other terms have to be taken into account, and the ionic sizes begin to play an important role. In fact, as one fixes  $\ell_B$  ( $\simeq 7.1$  Å in water at 25 °C) and the valences  $q_+$  and  $q_-$ , the only free parameters in the rhs of equation (73) are the ionic diameters  $d_+$  and  $d_-$ , which can be used in the theoretical predictions to fit the experimental values. This leads to effective equilibrium values of the ionic diameters (when in solution), which we call the “thermodynamic diameter”, in contrast to the bare diameter [1] (obtained through crystallographic methods) and the hydrodynamic diameter (obtained from mobility measurements [34]). By construction, it is this diameter which should be used in equilibrium situations if one wants to describe an electrolyte solution as a TCPHC, as for example computer simulations of electrolytes.



**Fig. 6.** Experimental [31,35] and fitted theoretical mean activity coefficient  $f_{\pm}$  for various salts as a function of the density  $\rho$  (in mole/liter), for various 1:1 salts, *viz.* HCl (triangles and dot-dashed line,  $d = 4.0$  Å), NaCl (stars and dotted line,  $d = 3.7$  Å) and KCl (diamonds and dashed line,  $d = 3.6$  Å). The full line denotes the Debye-Hückel limiting law, which scales with  $\rho^{1/2}$ . The fit was done using the data points shown (up to  $\rho = 0.02$  mole/liter).

### 3.3.1 Fitting assuming one mean diameter

We now show the fitting procedure assuming that  $d_+ = d_- = d$ , where  $d$  is the mean diameter. We also restrict here this procedure to 1:1 salts, where more experimental data is available at reasonably low densities (below  $\sim 0.05$  mole/liter). For asymmetric salts, this method demands precise data at the range below  $\sim 0.01$  mole/liter, where Debye-Hückel is often assumed to account for all effects and few experimental points are available.

The assumption of equal ionic sizes has been often used in the past to fit activities to modified Debye-Hückel theories [28] to account for the ionic sizes (as previously mentioned, the Debye-Hückel limiting law is insensitive to it). Since we force the two diameters to be equal, we only need the expansion for  $f_{\pm}$  up to linear order since this will determine uniquely the single unknown parameter. From the experimental data  $f_{\pm}^{\text{exp}}$  [31,35] we subtract the diameter-independent term and obtain the difference

$$\Delta f_{\pm} \equiv f_{\pm}^{\text{exp}} - \nu_{\text{DH}} \rho^{1/2} = \nu_1^{\text{exp}} \rho + O(\rho^{3/2}). \quad (74)$$

The fit to this function leads to the coefficient of the linear term  $\nu_1^{\text{exp}}$ , and that can be used to determine the diameter by solving the equation  $\nu_1(d) = \nu_1^{\text{exp}}$  (with  $d$  as unknown). In Figure 6 we show the experimental [31,35]  $f_{\pm}$  for HCl, NaCl and KCl and the theoretical results (up to linear order in  $\rho$ ) using, respectively, the diameters 4.0 Å, 3.7 Å and 3.6 Å. These values come from the fitting described above applied to the experimental data in the range  $\rho = 0.01$  to 0.05 mole/liter. The diameters obtained are very close to the ones obtained in reference [35] with a similar fitting, but using the DH theory with an approximate way to incorporate the ionic sizes.

The fact that  $d_{\text{HCl}} > d_{\text{NaCl}} > d_{\text{KCl}}$ , which is the opposite to the sequence of bare diameters, is a consequence of the existence of hydration shells around the ions which tend to be larger for smaller bare ion size [1]. Notice however that the values obtained here lie between the bare

diameters and the hydrated values available in the literature. This is not surprising since the effective hard-core size simultaneously reflects both the presence of the hydration shells and the “softness” of the outer water layer as two oppositely charged ions approach each other.

### 3.3.2 Fitting assuming two diameters

If we now assume that both  $d_+$  and  $d_-$  are unconstrained, we have to use one more term in the expansion of the activity coefficient and solve the coupled equations

$$\begin{aligned} \nu_1(d_+, d_-) &= \nu_1^{\text{exp}} \\ \nu_{3/2}(d_+, d_-) &= \nu_{3/2}^{\text{exp}} \end{aligned} \quad (75)$$

where the coefficients  $\nu_1^{\text{exp}}$  and  $\nu_{3/2}^{\text{exp}}$  are obtained by fitting the function

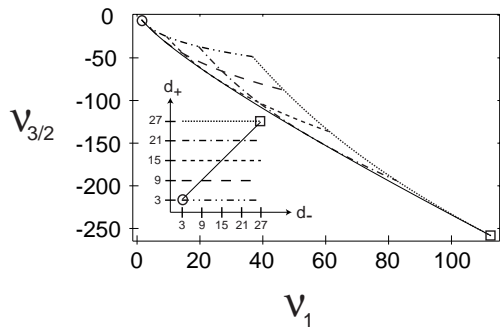
$$\begin{aligned} \Delta f_{\pm} &= f_{\pm}^{\text{exp}} - \nu_{\text{DH}} \rho^{1/2} - \nu_{1\log} \rho \ln \rho \\ &= \nu_1^{\text{exp}} \rho + \nu_{3/2}^{\text{exp}} \rho^{3/2} + O(\rho^{3/2} \ln \rho). \end{aligned} \quad (76)$$

For the actual fitting we divide by the density and obtain

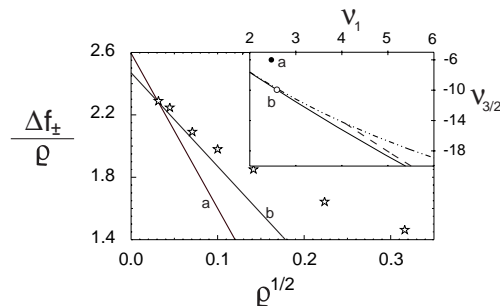
$$\frac{\Delta f_{\pm}}{\rho} = \nu_1^{\text{exp}} + \nu_{3/2}^{\text{exp}} \rho^{1/2}. \quad (77)$$

Note that we subtracted the term proportional to  $\rho \ln \rho$ , since like the DH term it is independent of the ionic diameters. What is interesting about this fitting procedure is that the two ionic diameters are independent parameters, that is, the effective sizes obtained through this fitting do not depend on the size of a “reference” ion, contrary to what happens when using crystallographic methods [36]. The method we show here can in principle lead to useful results, as long as the experimental data is accurate enough at very low densities, as we now demonstrate.

Figure 7 shows the mapping between the parameter space of ionic diameters and the two coefficients  $\nu_1$  and  $\nu_{3/2}$  (for a 1:1 salt in water at room temperature). The mapping is restricted to the range  $3 \text{ Å} < d_+ < 27 \text{ Å}$  and  $3 \text{ Å} < d_- < 27 \text{ Å}$ . This allows the “inverse mapping” between the  $(\nu_1; \nu_{3/2})$ -space and the  $(d_-; d_+)$ -space: any system with ionic diameters within the values above should have the coefficients  $\nu_1$  and  $\nu_{3/2}$  inside the region in the  $(\nu_1; \nu_{3/2})$ -space, as shown in Figure 7. One problem becomes obvious: the coefficient  $\nu_{3/2}$  is larger than the coefficient of the linear term,  $\nu_1$ , meaning that experimental data at very low concentrations are needed. As an example, Figure 8 shows  $\Delta f_{\pm}/\rho$  for NaCl [31] as a function of  $\rho^{1/2}$ , which should be asymptotically linear in the limit  $\rho \rightarrow 0$  and therefore give the coefficients  $\nu_1$  and  $\nu_{3/2}$  by a simple linear fit according to equation (77). The lines shown are two possible asymptotic linear fits to the experimental points (line a is given by  $\Delta f_{\pm}/\rho = 2.59 - 9.94 \rho^{1/2}$ , and line b by  $2.47 - 6.00 \rho^{1/2}$ ). The inset to Figure 8 gives the corresponding positions in the  $(\nu_1; \nu_{3/2})$ -space (*cf.* Fig. 7) for the two lines. Clearly, only one of the fits (line b) leads to reasonable values for the diameters (between 3 and 9 Å as seen from the position of point b in the



**Fig. 7.** Mapping ionic diameters of a 1:1 salt in water at room temperature into coefficients of the rescaled activity coefficient  $\Delta f_{\pm}$  defined in equation (77). The inset shows different paths in the  $(d_-; d_+)$ -space (with the ionic diameters between 3 Å and 27 Å) which are mapped into lines in the  $(\nu_1; \nu_{3/2})$ -space, as determined by equations (100–102). By knowing the values of  $\nu_1$  and  $\nu_{3/2}$  for a certain salt at very low densities, one can make the “inverse-mapping” and extract the values of  $d_-$  and  $d_+$ . The solid line, the symmetric case with  $d_+ = d_-$ , is the lower envelope of all other lines which are obtained by varying one diameter while keeping the other fixed.



**Fig. 8.** Fitting procedure (NaCl in water at room temperature) for two unconstrained diameters. Lines a and b correspond to two possible asymptotic fits of  $\Delta f_{\pm}/\rho$  as  $\rho \rightarrow 0$  (see text). From these lines one can extract  $\nu_1$  and  $\nu_{3/2}$ , which allows determination of the ionic diameters (inset, which shows a part of the mapping done in Fig. 7). Notice that while the two lines are reasonable asymptotic fits to the data, one (line b) leads to acceptable values for the ionic diameters,  $d_+ = 3.8$  Å and  $d_- = 5.4$  Å, while the other (line a) leads to values that lie outside the range 3 to 27 Å (which is not reasonable). This means that more precise data in the range  $\rho < 0.01$  mole/liter is needed to obtain the correct values for the diameters with this method. Experimental data from reference [31].

inset,  $d_+ = 3.8$  Å and  $d_- = 5.4$  Å as obtained by solving Eq. (75)), while the other fit (line a) leads to unreasonable values for the diameters (outside the range 3 to 27 Å). In other words, the asymptotic extrapolation of the experimental data is very sensitive to small errors, and in order to obtain the diameters with this method one needs more accurate data at very low densities (which to the best of our knowledge is not available in the literature). Although NaCl was used as example, the situation is identical for other salts.

## 4 Conclusions

Using field theoretic methods we obtained the exact low density (“virial”) expansion of the TCPHC up to order 5/2 in density. In its general form, the model can be applied to both electrolyte solutions and dilute colloidal suspensions (when the van der Waals forces are unimportant); the free energy derived here provides a unified way for handling both systems in the limit of low concentration. As the calculations show, the generalization to short-ranged potentials other than hard core is possible.

The behavior of the coefficients  $B_2$  and  $B_{5/2}$  suggests that the series is badly convergent, meaning that the inclusion of higher order terms does not necessarily extend the validity of the free energy to larger densities. We saw that the divergent behavior of these coefficients is related to the ionic pairing [14–16] which is favored as the coupling increases (this is not present *e.g.* in the OCPHC [10]). This also means that such a low-density expansion is not very useful to study the phase behavior [37] of ionic systems: in the situation of phase-separation between a very dilute phase and a dense phase, typically the average density in the dense phase already falls outside the range of validity of the low density expansion.

In applying our results to colloidal systems we concluded that at low density the counterion entropic contribution dominates over the electrostatic contribution due to the macroions; the latter contribution can be described, in this limit and up to this order, by an OCPHC corrected by effects due to the exclusion of counterions from the colloidal particles.

Finally, we used the theoretical results for the mean activity coefficient to fit experimental data and extract effective ionic sizes. In the simplest fitting, where we assumed the two ionic diameters to be equal, we obtained sizes that are reasonable and close to what one would expect from the results obtained by other methods. For the more interesting case where the two ionic sizes are taken as free parameters and determined independently, we showed that one would need more experimental data for the mean activity coefficient at very low densities (which, to the best of our knowledge, is not available in the literature) to obtain the correct values for the diameters. With the proper experimental data it would then be a simple matter to obtain the effective thermodynamic ionic sizes, which could serve as useful input for computer simulations of two-component-plasmas with hard-cores to model electrolyte solutions.

We thank N. Brilliantov and H. Löwen for useful discussions. AGM acknowledges the support from FCT through grant PRAXIS XXI/BD/13347/97 and DFG.

## Appendix

### Averages needed for $Z_1$ and $Z_2$

The following expressions were used to obtain equations (28, 29). In order to have more compact formulas

we use the Greek letters  $\alpha$  and  $\beta$  instead of  $+$  or  $-$ . For instance,  $\alpha q_\alpha$  means both  $+q_+$  and  $-q_-$

$$\langle h_\alpha(\mathbf{r}) \rangle = e^{q_\alpha^2 v_c(0)/2}, \quad (78)$$

$$\langle e^{-i\alpha q_\alpha \phi(\mathbf{r})} \rangle = e^{-q_\alpha^2 v_{\text{DH}}(0)/2}, \quad (79)$$

$$\langle h_\alpha(\mathbf{r}) h_\beta(\mathbf{r}') \rangle = e^{-\omega_{\alpha\beta}(\mathbf{r}-\mathbf{r}') + [q_\alpha^2 v_c(0) + q_\beta^2 v_c(0)]/2}, \quad (80)$$

$$\langle e^{-i\alpha q_\alpha \phi(\mathbf{r}) - i\beta q_\beta \phi(\mathbf{r}')} \rangle = e^{-[q_\alpha^2 v_{\text{DH}}(0) + q_\beta^2 v_{\text{DH}}(0)]/2} \times e^{-\alpha\beta q_\alpha q_\beta v_{\text{DH}}(\mathbf{r}-\mathbf{r}')} \quad (81)$$

$$\langle \phi^2(\mathbf{r}') e^{-i\alpha q_\alpha \phi(\mathbf{r})} \rangle = e^{-q_\alpha^2 v_{\text{DH}}(0)/2} \times [v_{\text{DH}}(0) - q_\alpha^2 v_{\text{DH}}^2(\mathbf{r}-\mathbf{r}')]. \quad (82)$$

The brackets  $\langle \dots \rangle$  denote averages over the fields  $\phi$  and  $\psi_\alpha$  where  $\omega^{-1}$  and  $v_{\text{DH}}^{-1}$  are the propagators.

### The coefficients in the grand-canonical free energy

We give here the explicit expressions for the coefficients  $\tilde{g}$ , equation (32). Using the function  $H(x)$ , defined in equation (49), the coefficients read

$$m_1 = \frac{\pi}{3} \frac{q_+^6 \ell_B^3}{d_+^3} \left\{ -H\left(\frac{q_+^2 \ell_B}{d_+}\right) + \ln\left(\frac{3d_+}{\ell_B}\right) \right\}, \quad (83)$$

$$m_2 = \frac{\pi}{3} \frac{q_-^6 \ell_B^3}{d_-^3} \left\{ -H\left(\frac{q_-^2 \ell_B}{d_-}\right) + \ln\left(\frac{3d_-}{\ell_B}\right) \right\}, \quad (84)$$

$$m_3 = \frac{2\pi}{3} \frac{q_+^3 q_-^3 \ell_B^3}{d_+^3} \left\{ H\left(\frac{2q_+ q_- \ell_B}{d_+ + d_-}\right) - \ln\left(\frac{3[d_+ + d_-]}{2\ell_B}\right) \right\}, \quad (85)$$

$$n_1 = \frac{\pi}{3} \frac{q_+^8 \ell_B^3}{d_+^3} \left\{ -\frac{5}{8} - 2H\left(\frac{q_+^2 \ell_B}{d_+}\right) + \ln\left(\frac{12d_+^2}{\ell_B^2}\right) \right\}, \quad (86)$$

$$n_2 = \frac{\pi}{3} \frac{q_-^8 \ell_B^3}{d_-^3} \left\{ -\frac{5}{8} - 2H\left(\frac{q_-^2 \ell_B}{d_-}\right) + \ln\left(\frac{12d_-^2}{\ell_B^2}\right) \right\}, \quad (87)$$

$$n_3 = \frac{2\pi}{3} \frac{q_+^3 q_-^3 \ell_B^3}{d_+^3} \left\{ \frac{5}{8} q_+ q_- + \frac{[q_+ - q_-]^2}{2} H\left(\frac{2q_+ q_- \ell_B}{d_+ + d_-}\right) + q_+ q_- \ln\left(\frac{2[d_+ + d_-]}{\ell_B}\right) - \frac{q_+^2 + q_-^2}{2} \ln\left(\frac{3[d_+ + d_-]}{2\ell_B}\right) \right\}, \quad (88)$$

$$p_1 = \frac{\pi}{3} \frac{q_+^6 \ell_B^3}{d_+^3}, \quad p_2 = \frac{\pi}{3} \frac{q_-^6 \ell_B^3}{d_-^3}, \quad p_3 = -\frac{2\pi}{3} \frac{q_+^3 q_-^3 \ell_B^3}{d_+^3}, \quad (89)$$

$$r_1 = \frac{2\pi}{3} \frac{q_+^8 \ell_B^3}{d_+^3}, \quad r_2 = \frac{2\pi}{3} \frac{q_-^8 \ell_B^3}{d_-^3}, \quad (90)$$

$$r_3 = \frac{2\pi}{3} \frac{q_+^3 q_-^3 \ell_B^3}{d_+^3} \left[ q_+ q_- - \frac{q_+^2 + q_-^2}{2} \right], \quad (91)$$

$$s_1 = \frac{q_+^4}{8}, \quad s_2 = \frac{q_-^4}{8}, \quad t_0 = \frac{d_+^3}{12\pi \ell_B^3}, \quad t_1 = \frac{q_+^6}{48}, \quad t_2 = \frac{q_-^6}{48}. \quad (92)$$

### The coefficients in the fugacity

In equation (39) we define the fugacity of negative ions  $\tilde{\lambda}_-$  as an expansion in terms of  $\tilde{\lambda}_+$ . The coefficients  $a_0$ ,  $a_1$ , etc. are determined by the condition of global electroneutrality. They are given by

$$a_0 = \frac{q_+}{q_-}, \quad (93)$$

$$a_1 = \frac{q_+^2}{q_-} [q_+^2 - q_-^2] \sqrt{\frac{\pi \ell_B^3}{d_+^3} \left[ 1 + \frac{q_-}{q_+} \right]}, \quad (94)$$

$$a_2 = \frac{q_+}{q_-^2} [-2m_2 q_+ + 2m_1 q_- + m_3 [q_1 - q_2]] + \frac{\pi}{6} \frac{q_+^2 \ell_B^3}{q_- d_+^3} [q_+ + q_-]^2 [7q_+^3 - 9q_+^2 q_- + 2q_-^3] + \frac{\pi}{3} \frac{q_+^2 \ell_B^3}{q_- d_+^3} [q_+^5 - q_+^3 q_-^2 + q_+^2 q_-^3 - q_-^5] \times \ln\left(\frac{4\pi \ell_B^3}{d_+^3} q_+ [q_+ + q_-]\right), \quad (95)$$

$$a_3 = \frac{\pi}{3} \frac{q_+^2 \ell_B^3}{q_- d_+^3} [q_+^5 - q_+^3 q_-^2 + q_+^2 q_-^3 - q_-^5], \quad (96)$$

$$a_4 = \frac{\ell_B^3/2}{24q_-^3} \sqrt{\pi q_+ (q_+ + q_-)} \times \left\{ 24n_1 [5q_+ q_-^2 - q_-^3] + 24n_2 [q_+^3 - 5q_+ q_-^2] + 72n_3 [q_+^2 q_- - q_+ q_-^2] + 24m_1 [q_+^2 q_-^3 - 3q_+ q_-^4] + 24m_2 [-4q_+^4 q_- - q_+^3 q_-^2 + 7q_+^2 q_-^3] + 12m_3 [2q_+^4 q_- - q_+^3 q_-^2 - 6q_+^2 q_-^3 + 5q_+ q_-^4] + \frac{\pi \ell_B^3}{d_+^3} q_+^2 q_-^2 [q_+ + q_-]^2 [26q_+^5 - 34q_+^4 q_- - 31q_+^3 q_-^2 + 67q_+^2 q_-^3 - 45q_+ q_-^4 + 17q_-^5] \right\} + \frac{\pi^{3/2}}{6} \frac{q_+^{5/2} \ell_B^{9/2}}{q_- d_+^{9/2}} [q_+ - q_-] [q_+ + q_-]^{5/2} \times [10q_+^4 - 11q_+^3 q_- + 13q_+^2 q_-^2 - 4q_+ q_-^3 + 3q_-^4] \times \ln\left(\frac{4\pi \ell_B^3}{d_+^3} q_+ [q_+ + q_-]\right), \quad (97)$$

$$a_5 = \frac{\pi^{3/2}}{6} \frac{q_+^{5/2} \ell_B^{9/2}}{q_- d_+^{9/2}} [q_+ - q_-] [q_+ + q_-]^{5/2} \times [10q_+^4 - 11q_+^3 q_- + 13q_+^2 q_-^2 - 4q_+ q_-^3 + 3q_-^4] \times \ln\left(\frac{4\pi \ell_B^3}{d_+^3} q_+ [q_+ + q_-]\right). \quad (98)$$

Notice that  $m_1, m_2$ , etc. were defined in equations (83–92).

### The coefficients in the mean activity coefficient

In equation (73) we obtained the low-density expansion of the mean activity coefficient of ionic solutions where the

ions have valences  $q_+$  and  $q_-$  and effective diameters  $d_+$  and  $d_-$  (in angströms). Defining  $\omega \equiv 6.022 \times 10^{-4} \ell_B^3 q_+^6 q_-$  (with  $\eta \equiv q_-/q_+$  and  $\xi \equiv d_-/d_+$ ), the coefficients in equation (73) read

$$\nu_{\text{DH}} = -\eta \sqrt{\pi\omega[1+\eta]}, \quad (99)$$

$$\begin{aligned} \nu_1 = & \frac{\pi\omega\eta}{6[1+\eta]} \left\{ -1 + 3\eta + 8\eta^2 + 3\eta^3 - \eta^4 + 4 \left[ H(\epsilon_+) \right. \right. \\ & + \ln(\epsilon_+) \left. \right] + 4\eta^4 \left[ H\left(\frac{\eta^2\epsilon_+}{\xi}\right) + \ln\left(\frac{\eta^2\epsilon_+}{\xi}\right) \right] \\ & - 8\eta^2 \left[ H\left(-\frac{2\eta\epsilon_+}{1+\xi}\right) + \ln\left(\frac{2\eta\epsilon_+}{1+\xi}\right) \right] \\ & + 8\eta^2[1-\eta^2] \ln(\eta) - [2-4\eta^2+2\eta^4] \\ & \times \ln(36\pi\omega[1+\eta]) \left. \right\}, \quad (100) \end{aligned}$$

$$\nu_{1\log} = -\frac{\pi\omega\eta}{3}[1-\eta]^2[1+\eta], \quad (101)$$

$$\begin{aligned} \nu_{3/2} = & \frac{\eta[\pi\omega]^{3/2}}{24\sqrt{1+\eta}} \left\{ 42 - 6\eta - 14\eta^2 + 68\eta^3 - 14\eta^4 - 6\eta^5 \right. \\ & + 42\eta^6 + 40 \left[ 1 - \frac{2\eta}{5} \right] \left[ H(\epsilon_+) + \ln(\epsilon_+) \right] \\ & + 40\eta^6 \left[ 1 - \frac{2\eta}{5\eta} \right] \left[ H\left(\frac{\eta^2\epsilon_+}{\xi}\right) + \ln\left(\frac{\eta^2\epsilon_+}{\xi}\right) \right] \\ & + 112\eta^3 \left[ H\left(-\frac{2\eta\epsilon_+}{1+\xi}\right) + \ln\left(\frac{2\eta\epsilon_+}{1+\xi}\right) \right] \\ & - 16\eta^3[7-2\eta^2+5\eta^3] \ln(\eta) \\ & + 8\eta[1-\eta^2]^2 \ln(36\pi\omega[1+\eta]) \\ & \left. - 20[1+\eta^3]^2 \ln(64\pi\omega[1+\eta]) \right\}, \quad (102) \end{aligned}$$

$$\begin{aligned} \nu_{3/2\log} = & -\frac{\eta[\pi\omega]^{3/2}}{6} \sqrt{1+\eta} \\ & \times [5-7\eta+7\eta^2+7\eta^3-7\eta^4+5\eta^5]. \quad (103) \end{aligned}$$

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